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## Structure Reports

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## Key indicators

Single-crystal X-ray study
$T=295 \mathrm{~K}$
Mean $\sigma(\mathrm{C}-\mathrm{C})=0.007 \AA$
$R$ factor $=0.058$
$w R$ factor $=0.101$
Data-to-parameter ratio $=12.2$
For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.
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## catena-Poly[[[aquabis(1 H-benzimidazole- $\kappa \mathrm{N}^{3}$ )-nickel(II)]- $\mu$-isophthalato- $\left.\kappa^{2} O: O^{\prime}\right]$ hemihydrate]

In the title polymeric complex, $\left[\mathrm{Ni}\left(\mathrm{C}_{8} \mathrm{H}_{4} \mathrm{O}_{4}\right)\left(\mathrm{C}_{7} \mathrm{H}_{6} \mathrm{~N}_{2}\right)_{2^{-}}\right.$ $\left.\left(\mathrm{H}_{2} \mathrm{O}\right)\right] \cdot 0.5 \mathrm{H}_{2} \mathrm{O}$, the $\mathrm{Ni}^{\mathrm{II}}$ atoms assume a distorted octahedral coordination geometry, formed by two isophthalate (ISOP) ions, two benzimidazole (BZIM) molecules and one water molecule. The ISOP dianions bridge the $\mathrm{Ni}^{\mathrm{II}}$ atoms through terminal carboxy groups to form zigzag polymeric chains, which are linked to each other via hydrogen bonds. $\pi-\pi$ stacking occurs between BZIM rings and between ISOP rings in the crystal structure.

## Comment

As part of an investigation of aromatic stacking in metal complexes (Liu et al., 2003; Su \& Xu, 2004), the title polymeric $\mathrm{Ni}^{\mathrm{II}}$ complex, (I), bridged by an isophthalate dianion (ISOP), incorporating benzimidazole (BZIM), has been prepared in the laboratory and its crystal structure is presented here.

(I)

The asymmetric unit of (I) contains two $\mathrm{Ni}^{\mathrm{II}}$ complex units and one uncoordinated water molecule. Each $\mathrm{Ni}^{\mathrm{II}}$ atom displays a similar coordination geometry, as shown in Fig. 1.


Figure 1
Part of the polymeric structure of (I), with $30 \%$ probability displacement ellipsoids. The dashed line indicates a hydrogen bond. [Symmetry code: (i) $1+x,-1+y, z$.]

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Figure 2
The molecular packing, showing polymeric complex chains. [Symmetry code: (ii) $1-x, 1-y, 1-z$.]

Two BZIM coordinate cis to an Ni atom, while two ISOP coordinate to the Ni atom in monodentate and chelate modes, respectively. One water molecule coordinates to the Ni atom to complete the distorted octahedral coordination geometry (Table 1). The uncoordinated water molecule is hydrogen bonded with carboxy groups coordinating to atom Ni1 (Table 2), but no uncoordinated water molecule links to the Ni2 complex unit.

Both ISOP groups have a nearly planar configuration, with a maximum dihedral angle of $10.0(6)^{\circ}$ between the C51benzene and C58-carboxy planes. The ISOP dianions bridge neighboring Ni atoms to form zigzag polymeric complex chains extending along the crystallographic [1六0] direction, as shown in Fig. 2. Adjacent chains are linked to each other via an extensive hydrogen-bonding network (Table 2).

An overlap between parallel BZIM rings and between parallel ISOP rings is observed in the crystal structure, as shown in Fig. 3. The distances between parallel N11-BZIM and N11-BZIM ${ }^{\mathrm{iii}}$ mean planes [symmetry code: (iii) $1-x$, $2-y, 1-z]$ and between N41-BZIM and N41-BZIM ${ }^{\text {iv }}$ mean planes [symmetry code: (iv) $2-x, 1-y,-z$ ] are 3.40 (3) and 3.53 (3) Å, respectively. The distance between parallel C61benzene and C61 ${ }^{\mathrm{v}}$-benzene rings [symmetry code: (v) $2-x$, $1-y, 1-z]$ is 3.28 (4) $\AA$. These distances clearly suggest the existence of $\pi-\pi$ stacking in the crystal. $\pi-\pi$ stacking also occurs between nearly parallel aromatic rings in the crystal structure (Fig. 4), as evidenced by the following data: the centroid-to-centroid distance of 3.394 (3) $\AA$ and dihedral angle of 10.4 (4) ${ }^{\circ}$ between N11-imidazole and N33-imidazole ${ }^{\text {ii }}$ rings [symmetry code: (ii) $1-x, 1-y, 1-z$ ]; the centroid-tocentroid distance of 3.526 (2) $\AA$ and dihedral angle of 2.0 (3) ${ }^{\circ}$ between C53-benzene and C63 ${ }^{\text {ii }}$-phenyl rings.

## Experimental

An ethanol solution ( 10 ml ) of BZIM $(0.24 \mathrm{~g}, 2 \mathrm{mmol})$ was mixed with an aqueous solution ( 4 ml ) of $\mathrm{NiCl}_{2} \cdot 6 \mathrm{H}_{2} \mathrm{O}(0.47 \mathrm{~g}, 2 \mathrm{mmol})$, and the mixture was refluxed for 25 min . An aqueous solution ( 8 ml ) containing isophthalic acid $(0.33 \mathrm{~g}, 2 \mathrm{mmol})$ and $\mathrm{NaOH}(0.16 \mathrm{~g}$, 4 mmol ) was then added to the above mixture, and the solution was refluxed for a further 3 h . After cooling to room temperature, the solution was filtered. Green single crystals were obtained from the filtrate after one week.




Figure 3
$\pi-\pi$ stacking between parallel aromatic rings. [Symmetry codes: (iii) $1-x, 2-y, 1-z$; (iv) $2-x, 1-y,-z$; (v) $2-x, 1-y, 1-z$.]

## Crystal data

$\left[\mathrm{Ni}\left(\mathrm{C}_{8} \mathrm{H}_{4} \mathrm{O}_{4}\right)\left(\mathrm{C}_{7} \mathrm{H}_{6} \mathrm{~N}_{2}\right)_{2-}\right.$ $\left.\left(\mathrm{H}_{2} \mathrm{O}\right)\right] \cdot 0.5 \mathrm{H}_{2} \mathrm{O}$
$M_{r}=486.12$
Triclinic, $P \overline{1}$
$a=12.1060(6) \AA$
$b=13.7927$ (8) $\AA$
$c=13.8994$ ( 9 ) $\AA$
$\alpha=60.813$ (3) ${ }^{\circ}$
$\beta=88.719$ (3) ${ }^{\circ}$
$\gamma=80.625(2)^{\circ}$
$V=1994.7$ (2) $\AA^{3}$
Data collection
Rigaku R-AXIS RAPID
diffractometer
$\omega$ scans
Absorption correction: multi-scan (ABSCOR; Higashi, 1995)
$T_{\text {min }}=0.81, T_{\text {max }}=0.90$
15366 measured reflections

## Refinement

Refinement on $F^{2}$
$R\left[F^{2}>2 \sigma\left(F^{2}\right)\right]=0.058$
$w R\left(F^{2}\right)=0.101$
$S=1.05$
7148 reflections
586 parameters
H -atom parameters constrained

## Table 1

Selected geometric parameters ( $\left({ }^{\circ},{ }^{\circ}\right)$.

| $\mathrm{Ni} 1-\mathrm{O} 1$ | $2.042(3)$ | $\mathrm{Ni} 2-\mathrm{O} 3^{\mathrm{i}}$ | $2.267(3)$ |
| :--- | :--- | :--- | :--- |
| $\mathrm{Ni} 1-\mathrm{O} 5$ | $2.089(3)$ | $\mathrm{Ni} 2-\mathrm{O} 4^{\mathrm{i}}$ | $2.098(3)$ |
| $\mathrm{Ni} 1-\mathrm{O} 6$ | $2.224(3)$ | $\mathrm{Ni} 2-\mathrm{O} 7$ | $2.039(3)$ |
| $\mathrm{Ni} 1-\mathrm{O} 11$ | $2.152(3)$ | $\mathrm{Ni} 2-\mathrm{O} 21$ | $2.134(3)$ |
| $\mathrm{Ni} 1-\mathrm{N} 13$ | $2.064(3)$ | $\mathrm{Ni} 2-\mathrm{N} 33$ | $2.075(3)$ |
| $\mathrm{Ni} 1-\mathrm{N} 23$ | $2.068(4)$ | $\mathrm{Ni} 2-\mathrm{N} 43$ | $2.096(4)$ |
|  |  |  |  |
| $\mathrm{O} 1-\mathrm{Ni} 1-\mathrm{O} 6$ | $112.62(11)$ | $\mathrm{O} 7-\mathrm{Ni} 2-\mathrm{O} 3^{\mathrm{i}}$ | $113.16(11)$ |

Symmetry code: (i) $1+x, y-1, z$.

Table 2
Hydrogen-bonding geometry $\left(\AA,{ }^{\circ}\right)$.

| $D-\mathrm{H} \cdots A$ | $D-\mathrm{H}$ | $\mathrm{H} \cdots A$ | $D \cdots A$ | $D-\mathrm{H} \cdots A$ |
| :---: | :---: | :---: | :---: | :---: |
| $\mathrm{O} 1 W-\mathrm{H} 1 A \cdots \mathrm{O} 5$ | 0.87 | 2.08 | 2.868 (4) | 151 |
| $\mathrm{O} 1 W-\mathrm{H} 1 B \cdots \mathrm{O} 2^{\text {ii }}$ | 0.88 | 2.32 | 3.004 (5) | 136 |
| $\mathrm{O} 11-\mathrm{H} 11 A \cdots \mathrm{O}^{\text {ii }}$ | 0.87 | 1.94 | 2.728 (4) | 150 |
| $\mathrm{O} 11-\mathrm{H} 11 \mathrm{~B} \cdots \mathrm{O}^{\text {ii }}{ }^{\text {a }}$ | 0.89 | 1.87 | 2.714 (4) | 158 |
| $\mathrm{O} 21-\mathrm{H} 21 A \cdots \mathrm{O} 3^{\text {ii }}$ | 0.93 | 1.96 | 2.788 (3) | 147 |
| $\mathrm{O} 21-\mathrm{H} 21 B \cdots \mathrm{O} 8^{\text {vi }}$ | 0.94 | 1.69 | 2.619 (5) | 172 |
| $\mathrm{N} 11-\mathrm{H} 11 \cdots \mathrm{O} 1^{\text {ii }}$ | 0.86 | 2.16 | 2.890 (4) | 142 |
| $\mathrm{N} 21-\mathrm{H} 21 \cdots \mathrm{O} 4^{\text {vii }}$ | 0.86 | 2.19 | 3.038 (5) | 168 |
| N31-H31 . ${ }^{\text {O }} 11^{\text {ii }}$ | 0.86 | 2.33 | 3.014 (5) | 137 |
| N41-H41...O1W ${ }^{\text {v }}$ | 0.86 | 2.25 | 2.958 (6) | 139 |

Symmetry codes: (ii) $1-x, 1-y, 1-z$; (v) $2-x, 1-y, 1-z$; (vi) $2-x,-y, 1-z$; (vii) $1+x, y, z$.

$$
\begin{aligned}
& Z=4 \\
& D_{x}=1.619 \mathrm{Mg} \mathrm{~m}^{-3}
\end{aligned}
$$

Mo $K \alpha$ radiation
Cell parameters from 8095 reflections
$\theta=2.2-24.0^{\circ}$
$\mu=1.02 \mathrm{~mm}^{-1}$
$T=295$ (2) K
Plate, green
$0.20 \times 0.15 \times 0.10 \mathrm{~mm}$

7148 independent reflections
4851 reflections with $I>2 \sigma(I)$
$R_{\text {int }}=0.068$
$\theta_{\text {max }}=25.2^{\circ}$
$h=-14 \rightarrow 14$
$k=-16 \rightarrow 16$
$l=-16 \rightarrow 16$

$$
\begin{aligned}
& w= 1 /\left[\sigma^{2}\left(F_{o}{ }^{2}\right)+(0.0347 P)^{2}\right. \\
&+1.5337 P] \\
& \text { where } P=\left(F_{o}{ }^{2}+2 F_{c}{ }^{2}\right) / 3 \\
&(\Delta / \sigma)_{\max }=0.001 \\
& \Delta \rho_{\max }=0.34 \mathrm{e} \AA^{-3} \\
& \Delta \rho_{\min }=-0.34 \mathrm{e} \AA^{-3}
\end{aligned}
$$



Figure 4
$\pi-\pi$ stacking between nearly parallel aromatic rings. [Symmetry code: (ii) $1-x, 1-y, 1-z$.]

Water H atoms were located in a difference Fourier map and included in the refinement with fixed positional ( $\mathrm{O}-\mathrm{H}=0.869-$ $0.939 \AA$ ) and isotropic displacement parameters $\left(0.05 \AA^{2}\right)$. All other H atoms were placed in calculated positions ( $\mathrm{C}-\mathrm{H}=0.93 \AA$ and $\mathrm{N}-$ $\mathrm{H}=0.86 \AA$ ) and refined as riding, with $U_{\text {iso }}(\mathrm{H})=1.2 U_{\text {eq }}$ (carrier atom).

Data collection: PROCESS-AUTO (Rigaku Corporation, 1998); cell refinement: PROCESS-AUTO; data reduction: CrystalStructure (Rigaku/MSC and Rigaku Corporation, 2002); program(s) used to solve structure: SIR92 (Altomare et al., 1993); program(s) used to refine structure: SHELXL97 (Sheldrick, 1997); molecular graphics: ORTEP-3 for Windows (Farrugia, 1997); software used to prepare material for publication: WinGX (Farrugia, 1999).

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